# Federal State Budgetary Educational Institution of Higher Education "Dagestan State Medical University" Health Ministry of the Russian Federation

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Acting vice-Rector for Academic Affairs,

Professor R.M.Ragimov

01 july 2022. Parumy

### WORKING PROGRAMM

### on "BIOCHEMISTRY"

Discipline index – B1.O.23.

Specialty - 31.05.01 « General Medicine »

Level of higher education - specialty

Graduate qualifications - medical doctor

Faculty - Medical

The department - General and biological chemistry

Form of training - full-time

Course - 2

Semester - 3, 4

**Total labor input** (in credit units / hours) – 7/252

lectures - 44 hours

practical classes - 88 hours

laboratory classes -20 hours

independent work -64 hours

exam - 4 semester - 36 hours

The work program of the discipline "Biochemistry" is developed on the basis of the workplan of BPEP HE in the specialty 05.31.01. "General Medicine" (higher education level - specialty), approved by the Academic Council of the Federal State Budget Educational Institution of Higher Medical Education of the Dagestan State Medical University of the Health Ministry of the Russian Federation, Minutes No. 1 dated 08/29/2019, in accordance with the Federal State Educational Standards of Higher Education in the direction of preparation 05.31.01. "General Medicine" (higher education level - specialty), approved by order No. 95 of the Education and Science Ministry of the Russian Federation dated 12.08.2020.

The work program of the discipline was approved at a meeting of the Department of General and Biological Chemistry from 24.06.2022 Protocol No. 11.

Head of Department F.

prof. Nagiev E.R.

Work program agreed:

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V.R. Musaeva

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#### 1. PURPOSE AND OBJECTIVES OF DISCIPLINE MASTERTING

**PURPOSE** – to form knowledge about the **molecular mechanisms** of the physiological functions of the human body and their disorders in pathological conditions, about the basic laws of metabolic processes that determine the state of health and human adaptation to changes in the

conditions of the external and internal environment; substantiate biochemical mechanisms for the prevention and treatment of diseases, and biochemical methods for diagnosing diseases and monitoring the effectiveness of treatment.

#### **OBJECTIVES:**

- To study the chemical structure of organic substances and their metabolism in a healthy person;
- familiarization of students with the structure, properties and functions of basic biomolecules.
- to study of the pathways of the metabolism of nucleic acids, proteins, carbohydrates and lipids and their relationships.
- to study of the stages of energy metabolism, methods of storage and consumption of metabolic fuel by cells.
- formation of ideas about the basic principles of regulation and their mechanisms.
- •to show by examples the pathogenesis of diseases as a result of damage to biochemical mechanisms;
- to teach students the biochemical diagnosis of diseases of the digestive, cardiovascular and excretory systems of the body;
- to teach students to work with literature, that is, find and understand information on biochemistry when it becomes necessary, and use this information to solve medical problems.

#### 2. LIST OF PLANNED STUDY RESULTS

Competencies formed in the process of studying a discipline:

Category	Code and name of general	Code and name of the achievement indicator of general
(group) of	professional competence	professional competence
general		
profession		
al		
competenc		
ies		
	<b>GPC -5.</b> One is ble to assess	
	morphofunctional,	
	physiological conditions and	

pathological processes in the	
human body to solve	
professional problems.	
	ID-1 GPC-5
	<b>To know</b> functional systems, their regulation and self-
	regulation under the influence of environmental factors in
	normal and pathological processes
	<b>To be able</b> to evaluate the results of clinical, laboratory
	and functional diagnostics in solving professional
	problems.
	To possess algorithm of laboratory and functional
	diagnostics in solving professional problems,
	<b>ID -2 GPC-5</b> One can: interpret the data of the main
	physical-chemical, mathematical and natural-science
	research methods in solving professional problems
	<b>To know:</b> the chemical and biological essence of the
	processes occurring in a living organism at the molecular
	and cellular levels; structure and biochemical properties of
	the main classes of biologically important compounds, the
	main metabolic pathways of their transformations; the role
	of cell membranes and their transport systems in
	metabolism in the body;
	<b>To be able:</b> to evaluate and interpret the results of the
	most common laboratory diagnostic methods to identify
	pathological processes;
	- substantiate the nature of the pathological process and its
	clinical manifestations, the principles of pathogenic
	therapy of the most common diseases.
	<b>To possess</b> the ability to assess physiological conditions
	and pathological processes in the human body based on
	the results of a laboratory examination in solving
	professional problems.

#### 3. PLACE OF EDUCATIONAL DISCIPLINE IN THE STRUCTURE OF OOP:

The discipline "Biochemistry" refers to the base part of block 1 of discipline B1. 0. 23. in the specialty 05.31.01. "General Medicine."

The basic knowledge necessary to study the discipline is formed:

- through the study of the following disciplines - biology, chemistry; anatomy; histology, embryology, cytology; normal physiology.

The discipline "Biochemistry" is the previous one for the study of disciplines:

- > pathophysiology;
- ➤ pharmacology;
- > microbiology, virology;
- ➤ immunology;
- > professional disciplines.

To study the discipline "Biological chemistry" the following knowledge, skills and abilities formed by previous disciplines are required.

#### Bioorganic chemistry:

- ➤ **Knowledge** To know the structure of the most important organic compounds that make up human organs and tissues (in particular, DNA, RNA, proteins, etc.);
- > Skills To be able to write the formulas that make up the components of the human body's macromolecules (amino acids, nucleotides, etc.), indicate what chemical bonds they connect to each other;
- ➤ Attainments theoretical skills explaining the development mechanisms of some pathological processes.

#### Biology:

- ➤ **Knowledge** To know the functions of the most important organic compounds that make up human organs and tissues (in particular, DNA, RNA, proteins, etc.);
- ➤ **Skills** To be able to analyze the possible connections between the origin of life on Earth and the structure and functions of proteins and nucleic acids;
- ➤ **Attainments** theoretical skills explaining the mechanisms of development of some pathological processes in a living organism.

Sections of the discipline "Biochemistry"

	Sections of the discipline Biochemistry			
No	Name of sections			
1	The structure and function of proteins. Enzymes			
2	Vitamins and hormones			
3	Bioenergy and biooxidation			
4	Carbohydrate metabolism.			
5	Lipid metabolism.			
6	The metabolism of simple and complex proteins.			
7	Regulation of metabolism.			
	Biochemistry of individual organs and tissues			
8	The biosynthesis of nucleic acids and proteins.			
	The basics of molecular genetics			

## The list of disciplines, which mastering by students is necessary for the study of biological chemistry.

№	Discipline	Sections
2.	Bioorganic chemistry	The structure and reactivity of carbohydrates, lipids, peptides and proteins, nucleic acids. General idea of biologically active substances; chemistry of some vitamins and hormones
3.	Physiology	"Cell Physiology"; "Physiology of metabolism"; "Physiology of digestion"; "Endocrine glands"; "Blood"; "Physiology of water-salt metabolism"

	Physiology	"The structure of cell organization, chemistry, functions of organelles of the cell"; "Levels of organization of living matter"; "Assimilation and dissimilation, their unity and opposite. Features of the structure of proteins. " "The concept of genetic material and its properties. The molecular basis of heredity. The problems of genetic engineering, the concept of hereditary diseases."
5.	Physics	Biomechanics "(centrifugation, its use in the practice of biomedical research." Oscillations and waves. Acoustics "(action of ultrasound on matter." Thermodynamics of biological systems "(The body as an open system. Energy balance. Forms of energy in a living cell. Calorimeter.) "Optics." "(Concentration colorimetry. Colorimetry)." Bioelectric potentials "Membrane potential and its ionic nature.
6.	Anatomy	«Organogenesis "Age-related anatomy" "Comparative anatomy" "Achievements of anatomy in the development of the doctrine of man"

# 3. THE CAPACITY OF EDUCATIONAL DISCIPLINE AND TYPES OF CONTACT WORK

The total complexity of the discipline is \_\_7\_ credits.

Type of study			Number of hours per	
		Total hours	semester	
	, , , , , , , , , , , , , , , , , , ,		3	4
Classwork lessons (total), including:		152	100	52
Lectures (L)		44	28	16
Practical classes (PC)		88	60	28
Laboratory classes (LC)		20	12	8
Student's self-study (SSS) (total)		64	44	20
Type of intermediate certification (test, exam)		36		36
hours.		252	144	108
Total labor input	credit unit	7	4	3

### 5. STRUCTURE AND CONTENT OF EDUCATIONAL DISCIPLINE

# 5.1 Sections of academic discipline and competencies that should be mastered in their study

Nº	Section Content	The name of the discipline section	Controlle d competenc ies
1.	Biological functions of proteins. Amino acids as structural elements of a protein molecule. Structural organization of proteins. Physico-chemical properties of proteins. Reversible and irreversible protein precipitation reactions.	The physiological role and biological functions of proteins. The primary structure of proteins and its informational role. Secondary and tertiary structures of proteins. Protein conformation: formation stages, features of the influence of environmental conditions. Conformational lability of proteins. Features of the quaternary structure of the protein. The structure and functions of oligomeric proteins on the example of hemoglobin in comparison with myoglobin.  The molecular weight of proteins. Size and shape of protein molecules. Globular and fibrillar proteins. Physico-chemical properties of the protein: solubility, ionization, hydration, precipitation of proteins from solutions. Protein denaturation, reversibility of denaturation. Factors causing denaturation. Protein precipitation reactions.	<b>GPC</b> -5
2	Classification of proteins. Simple and complex proteins. Chromoproteins. Hemoglobin, chemical nature, structure, role	Classification of proteins. Simple proteins - albumin, globulin, protamine, histone, structure and biological role. Complex proteins, their representatives. Phosphoproteins, nucleoproteins, Chromoproteins, their representatives. Hemoglobin, chemical nature, structure, role. Derivatives of hemoglobin. Hemoglobinopathies.	GPC -5

3.	General characteristics and properties of enzymes. The chemical nature of enzymes.  The structure of enzymes. The mechanism of action of enzymes. Features of enzymatic catalysis.	General characteristics and basic properties of enzymes. Evidence of the protein nature of enzymes. The specificity of the action of enzymes. Types of specificity. Organ-specific enzymes. Isoenzymes, determination of the isoenzyme spectrum of enzymes in the clinic.  Enzymes as simple and complex proteins. Cofactors: prosthetic groups of enzymes, coenzymes, metal ions. Cofactors of enzymes, characteristics of the main coenzymes and their functions. The mechanism of action of enzymes. Enzyme-substrate complexes and the mechanism of their formation. The main parameters characterizing the dependence of the rate of the enzymatic reaction on the concentration of the substrate. Active and allosteric enzyme centers.	<i>GPC</i> -5
4.	activity.	tivators and inhibitors of enzymes. Types of inhibition. Regulation of enzyme activity. zyme inhibitors and their use as therapeutic agents. Classification and nomenclature of enzymes. Brief description of individual classes of enzymes. The concept of enzymopathies. Enzymes are medicines. Principles of enzyme diagnostics and enzyme therapy	<i>GPC</i> -5
5.	Some issues of nutritional biochemistry. Vitamins, hypovitaminosis and vitamin deficiency. Vitamin classification. Fat-soluble vitamins A, D, E, K. Ascorbic acid - vitamin C, biorol. Characterization of water soluble vitamins. Coenzyme role of vitamins	Introduction to metabolism. Biochemistry of nutrition. Ideas about interchangeable and irreplaceable components of food. The biological role of vitamins. Hypo-, hyper- and vitamin deficiencies. Anti-vitamins. Vitamin classification. Fat-soluble vitamins A, D, E, K, bio-role. Ascorbic acid is vitamin C. The role of vitamin C in collagen synthesis. The mechanism of development of scurvy (scorbut).  The biological role of water-soluble vitamins B1, B2, B6, B12, PP, P (rutin), N (biotin). Pantothenic and folic acid. Coenzyme role of water soluble vitamins. The participation of vitamins in the metabolism. Hypovitaminosis associated with a lack of water-soluble vitamins. The biological role of vitamin-like substances	<b>GPC</b> -5
6.	General characteristics of hormones. The mechanisms of action of hormones. Hormones of the hypothalamus, pituitary,	The mechanism of action of steroid and protein hormones. Membrane-intracellular and cytosolic mechanisms of hormone action. cAMP - as an intermediary between hormones and intracellular regulatory mechanisms. The role of adenylate cyclase and phosphodiesterase, protein kinases. Ca2	GPC -5

7.	thyroid and parathyroid glands.	+, cGMP as secondary intermediaries. Guanylate cyclase, the role of nitric oxide NO in the formation of cGMP. Characterization of the hormones of the hypothalamus, pituitary, thyroid and parathyroid glands.  Chemical structure, biological effect and	<b>GPC</b> -5
	Hormones of the brain and adrenal cortex. Hormones of the pancreas and gonads.	participation in the metabolism of hormones of the cortical and brain layer of the adrenal glands; hormones of the pancreas and gonads. The development of pathological conditions with hypoand hyperfunction of these endocrine glands. Prostaglandins and their derivatives biological role and effect on metabolism.	
8.	Bioenergy General patterns of metabolism. The formation and storage of energy in the cell.  Macroergic compounds.	General patterns of metabolism in a living organism. The main stages of the catabolism of substances and their relationships. Representations of specific and general ways of catabolism. The catabolism of nutrients (carbohydrates, fats, proteins) is the main source of energy necessary for vital processes. Energy-rich compounds as universal energy keepers in biological objects. Structural features and properties of macroergic compounds. Endergonic and exergonic reactions in a living cell; macroergic compounds. ADP-ATP cycle.  ATP and its analogues. Creatine phosphate and its role in bioenergy.	GPC -5
9.	Biooxidation Modern representations of biological oxidation. Mitochondrial electron transfer chain. Oxidative phosphorylation.	Features of oxidative processes in living tissues. Structures suitable for dehydrogenation. Dehydrogenation of substrates and oxidation of hydrogen to form water (tissue respiration) as an energy source for ATP synthesis. Thermoregulatory function of tissue respiration. Regulation of tissue respiration intensity by endogenous and exogenous substances. The structure of the main respiratory chain. The role of NAD, NADF, FAD and FMN. Coenzyme Q and cytochromes.  Biological oxidation and oxidative phosphorylation. P / O phosphorylation coefficient. Chemiosmotic theory of oxidative phosphorylation. Dissociation of oxidation and phosphorylation. Free breathing. Conjugation and uncoupling factors. Respiratory control. Substrate phosphorylation. Microsomal oxidation is a biological role.	GPC -5

10			CDC 5
10.	Assimilation of dietary carbohydrates. Glycogen exchange. Regulation of the synthesis and breakdown of glycogen Glucose catabolism.	Essential carbohydrate foods. Digestion. Glucose as a major metabolite of carbohydrate metabolism. The mechanism of transmembrane transfer of glucose and other monosaccharides to cells. Glycogen is a backup form of glucose. The structure, properties and distribution of glycogen. Glycogen biosynthesis and breakdown (mobilization) are processes that maintain a constant glucose level in the blood. Differences in glycogen mobilization in the liver and muscles. Regulation of the synthesis and breakdown of glycogen by hormones.  Aerobic breakdown is the main pathway for glucose catabolism in humans. Aerobic glycolysis as a glucose-specific pathway of catabolism. Energy effect of aerobic glycolysis and aerobic breakdown of glucose. Anaerobic decomposition (anaerobic glycolysis). The difference between the final proton acceptors during aerobic and anaerobic glycolysis. Regeneration of NAD + as a reaction that ensures the continuous course of the glycolytic process in tissues with a limited supply of oxygen or the absence of mitochondria in the cells. Regulation of glucose catabolism.	GPC -5
11	The final stage of the catabolism of nutrients. The tricarboxylic acid cycle.	The specific and general path of catabolism. Oxidative decarboxylation of pyruvic acid. Pyruvate dehydrogenase complex. The tricarboxylic acid cycle is the main source of tissue respiration substrates. The relationship of reactions of the common pathway of catabolism and CPE. Energy balance of aerobic glucose oxidation. Mechanisms of regulation of the general pathway of catabolism. Hypoenergetic conditions.	GPC -5
12.	Glucose synthesis (gluconeogenesis). The pentose phosphate pathway for glucose conversion. Regulation of blood glucose is normal, hyperhypoglucosemia in pathological conditions.	Glucose biosynthesis (gluconeogenesis) from non-carbohydrate substances. Substrates of gluconeogenesis in various physiological conditions: during fasting and during physical exertion. Routes of exchange of lactate (measles cycle). Regulation of glycolysis and gluconeogenesis. The role of insulin and glucagon. The importance of glycolysis in the liver for fat synthesis. Regulation of blood glucose in various physiological conditions of the body. The pentose phosphate pathway for glucose conversion. Distribution and physiological significance of the process.	GPC -5

3.	Assimilation of food lipids. Lipid transport by chylomicrons. Fat mobilization, β- oxidation of fatty acids. Metabolism of ketone bodies. Biosynthesis and functions of eicosanoids	The structure and functions of human tissue lipids, essential fatty acids. Digestion, absorption and transport of blood fats and possible violations of these processes: steatorrhea, hyperchilomicronemia. The function of lipoprotein lipase. Mobilization of fat in adipose tissue. The role of insulin, glucagon, adrenaline in the regulation of fat metabolism. $\beta$ – oxidation of fatty acids, its regulation. The biosynthesis and oxidation of ketone bodies. The role of fatty acids and ketone bodies as sources of energy during physical work, fasting, diabetes. Eicosanoids, biological effects. The use in dentistry of drugs that suppress the synthesis of eicosanoids.	<i>GPC</i> -5
	The biosynthesis of fatty acids and fats. Hormonal regulation.	Stages of fatty acid biosynthesis, synthesis of fats from carbohydrates in the liver, packaging in VLDL and transport. Deposition of fat in adipose tissue. The role of insulin in the regulation of fatty acid and fat synthesis.	GPC -5
14	Cholesterol metabolism, regulation of the process. Biosynthesis and function of bile acids. Hypercholesterolemia. Cell structure membranes and their role in metabolism and energy. The role of membranes in transmembrane signaling. Lipid peroxidation and antioxidant systems.	The functions of cholesterol, the stages of its biosynthesis and regulation. The role of lipoproteins in cholesterol transport. Synthesis and conjugation of bile acids, enterohepatic circulation. Hypercholesterolemia, biochemical basis for the development of atherosclerosis and its treatment. The role of ω-3 acids in the prevention of atherosclerosis complications. Gallstone disease and principles of its treatment. The main cell membranes and their functions. The lipid composition of the membranes is phospholipids, glycolipids, cholesterol. Mechanisms for the transfer of substances through membranes. The main components and stages of transmembrane signaling of hormones, mediators, cytokines, eicosanoids.  Lipid peroxidation (LPO) and its mechanism. Membrane damage due to LPO activation. The mechanism of protection against the toxic effects of oxygen: superoxide dismutase, catalase, glutathione peroxidase	GPC -5
15	Nitrogen balance. Digestion and absorption of proteins in the gastrointestinal tract. Decay of proteins in the intestines. Paired connections.	Digestion of proteins, absorption of amino acids. Peptidases of the stomach and pancreas. Essential and essential amino acids. Diagnostic value of biochemical analysis of gastric and duodenal juice. Decay of proteins in the intestines. The role of the liver in the neutralization of toxic substances and the formation of paired compounds.	GPC -5
	Common pathways of amino acid catabolism. Trans and deamination as an	The role of pyridoxalphosphate in the metabolism of amino acids. Transamination and deamination of amino acids. The biological significance of these	GPC -5

intermediate exchange of	processes. Oxidative deamination is the main type	
amino acids.	of deamination in human tissues. Glutamate	
william were as	dehydrogenase. The clinical significance of serum	1
	transaminases. Indirect zeamination of amino acids.	1
The exchange of individual		GPC -5
amino acids. The formation	of biogenic amines. Exchange of serine and glycine.	G1 C -3
and inactivation of biogenic	The role of H4 folate. The mechanism of action of	1
amines. Methionine and its	sulfa drugs. Methionine and transmethylation	1
participation in	reactions. The synthesis of creatine and its	1
transmethylation processes.	importance for providing energy to muscle work.	1
Inherited disorders of amino	The exchange of phenylalanine and tyrosine in	1
acid metabolism.	different tissues. Synthesis of catecholamines and	1
Education, causes of toxicity	their biological role. The final products of nitrogen	1
and neutralization of	metabolism are ammonium salts and urea. The role	1
ammonia. Urea synthesis	of glutamine and alanine in the neutralization and	1
Hyperammonemia.	transport of ammonia. Urea synthesis in the liver.	1
J.F. vi	Violations of the synthesis and elimination of urea,	1
	as the main cause of different types of	1
	hyperammonemia. Use of nitrogen-free amino acid	1
	residues	1
		1
		1

The exchange of complex proteins.

Transformation of nucleoproteins.

Biosynthesis of DNA and RNA. Repair errors and DNA damage.

Protein biosynthesis translation. Inhibitors of matrix biosynthesis. Mechanisms of genetic variation and protein polymorphism.

Regulation of protein synthesis. Molecular mutations.

The breakdown of nucleoproteins and nucleic acids. Ways of synthesis of purine and pyrimidine nucleotides, enzymes, regulation. Catabolism of purine and pyrimidine nucleotides. Pathology of purine nucleotide metabolism: gout. The structure and functions of nucleic acids. Features of the structure of DNA. DNA replication. The structure and functions of DNA and various types of RNA. DNA synthesis, providing the transmission of genetic traits from generation to generation. The association of replication with the cell cycle. DNA repair is the basis of genome stability.

**GPC** -5

Biological code as a way of translating a four-digit nucleotide record into an amino acid sequence. Protein synthesizing system. The sequence of events during the formation of the polypeptide chain on the ribosome. Amino Acid Activation. Broadcast and its mechanism. Stages of protein synthesis. Initiation, Elongation and Termination of Protein Synthesis. Postsynthetic changes and the formation of functionally active proteins.

Regulation of protein biosynthesis. The concept of operon and regulation at the transcription level. Induction and its mechanisms. Post-translational modifications of proteins. Inhibitors of matrix syntheses. Molecular mutations and recombinations as a source of genetic variation. Inherited diseases. The use of DNA technology in medicine.

Polymerase chain reaction and PCR diagnostics.

17 Changes in hormonal status and metabolism during fasting and diabetes.

Hormonal regulation of water-salt metabolism and calcium metabolism.

Regulation of energy metabolism, the role of insulin contrainsular hormones homeostasis. The role of insulin and glucagon in the metabolism regulation of energy postabsorption period and during fasting. Changes in hormonal status and metabolism in diabetes. Diabetic coma. The pathogenesis complications of diabetes mellitus (macro- and microangiopathies, nephropathy, retinopathy, cataracts, caries, periodontal disease).

Regulation of water-salt metabolism. The structure and functions of aldosterone, vasopressin, atrial natriuretic factor (PNF). The renin-angiotensin-aldosterone system. Biochemical mechanisms of the occurrence of renal hypertension, edema, xerostomia. The role of hormones in the regulation of calcium and phosphate metabolism (parathyroid hormone, calcitonin and calcitriol). The structure, biosynthesis and mechanism of action of calcitriol. Causes and manifestations of rickets, hypo- and hyperparathyroidism

The chemical composition of the blood. Proteins of blood plasma. Red blood cell metabolism.

Blood Buffer enzymes. Organic systems. and inorganic components of the blood. Hem metabolism and formation the of bile pigments. Jaundice Biochemistry of connective Collagen, elastin, tissue. proteoglycans, their role.

The chemical composition of the blood. Proteins of blood plasma. Albumins, globulins, fibrinogen. Proteins of the acute phase, their determination for the purpose of diagnosis. Features of the structure of red blood cells. Glucose metabolism and neutralization of reactive oxygen species in red blood cells. Hemoglobinopathies.

Enzymes of blood plasma, their definition for the diagnosis of various diseases. Non-protein nitrogenous components of the blood. Nitrogen-free organic components of the blood. Azotemia. Inorganic components of the blood. Blood buffer systems and acid-base balance.

The structure and biosynthesis of heme, regulation. Disorders of heme biosynthesis - porphyria. Iron metabolism: absorption, transport, entry into cells. Violations of iron metabolism. Hem catabolism. Bilirubin metabolism. Jaundice and their differential diagnosis. Hereditary metabolic disorders of bilirubin.

Features of the synthesis of intracellular and extracellular post-translational modifications of the proteins of the intercellular matrix. The structure and functions of glycosaminoglycans. Hereditary and acquired disorders of protein metabolism of connective tissue.

Non-collagenous bone proteins: osteonectin, osteocalcin, osteopontin; features of their structure and metabolism. Changes in connective tissue with

**GPC-5** 

**GPC** -5

	aging.		

### 5.1. Sections of the discipline and labor intensity by types of educational work

Section No	Name of the discipline section	Types of educational work  classwork extra		al work, hour.	Total hour.	
74≅				extracurricular	nour.	
		L	PC	LC		
					*SSS	
1.	The structure and functions of proteins Enzymes.	6	16	4	8	34
2.	vitamins Hormones	8	20	4	18	50
3	Bioenergetics. Biooxidation	6	10	2	8	26
4	Carbohydrate metabolism	8	14	2	10	34
	TOTAL for the 3 semester	28	60	12	44	144
5	lipid metabolism. The structure of biological membranes. Lipid peroxidation and antioxidant systems.	4	8	2	4	18
6	Metabolism of simple and complex proteins. Amino acid metabolism.	4	6	2	6	18
7	Biosynthesis of nucleic acids and proteins (matrix syntheses).	4	6	2	4	16
8	The regulation of metabolism. Biochemistry of individual organs and tissues (Biochemistry of the liver, blood, urine, connective tissue).	4	8	2	6	20

TOTAL for the 4 semester	16	28	8	20	72
TOTAL FOR THE YEAR	44	88	20	64	216

**5.3** Thematic plan of lectures

/	Section of	5.3 Thematic plan of lectures	3	4
п/ №	discipline	Thematics of lectures	term	term
712	Proteins and	Lecture №1 Structural organization of proteins. Features of	term	term
		the functioning of oligomeric proteins.	2	
	enzymes		2	
1		<b>Lecture №2</b> Physico-Chemical Properties of Proteins	2	
•		Lecture №3 Features of enzymatic catalysis. Regulation	2	
		of enzyme activity. The role of vitamins enzyme inhibitors.	2	
		The use of enzymes in medicine	_	
	Vitamins and	<b>Lecture №</b> Water soluble and fat soluble vitamins.		
	Hormones	<b>Lecture №5</b> Coenzyme form of vitamins.	2	
2		Lecture №6 Hormones, mechanism of action, structure.	2 2	
		Lecture №7 Hormones of the hypothalamus, pituitary	2	
		gland, pancreas, adrenal glands.		
	Bioenergy and	Lecture №8 Common pathway of catabolism.		
	biooxidation	Lecture №9 tissue respiration. Mitochondrial electron	2	
3		transport chain.		
		<b>Lecture №10</b> Oxidative phosphorylation of ADP.	2	
		Mitchell's chemioosmotic theory		
	Carbohydrate	<b>Lecture №11</b> Assimilation of dietary carbohydrates.		
	metabolism.	Synthesis and mobilization of glycogen, regulation of		
		processes.		
		Lecture №22 Aerobic and anaerobic glycolysis. Energy	2	
4		effect of processes.		
•		<b>Lecture №13</b> Oxidative decarboxylation of pyruvate and		
		tricarboxylic acid cycle.	2	
		<b>Lecture №14</b> Gluconeogenesis is the synthesis of glucose		
		from non-carbohydrate substances. Hormonal regulation of		
		the process		
		TOTAL	28 h	
	lipid metabolism	<b>Lecture №15</b> Assimilation of dietary lipids. Transport of		2
		lipids by chylomicrons. Biosynthesis of fatty acids and fats.		
_		Hormonal regulation.		
5		<b>Lecture №16</b> TAG mobilization. Process regulation. B-		2
		oxidation. Exchange of ketone bodies. Eicosanoids.		
		cholesterol exchange. Hypercholesterolemia. Synthesis of		
	D	bile acids. biological membranes. lipid peroxidation.		
	Protein	Lecture №17 Digestion of proteins. Replaceable and		2
	metabolism	irreplaceable amino acids. Trans- and deamination of amino		
6		acids.  Lecture №18 Neutralization of ammonia in tissues.		2
6				
		ornithine cycle. Synthesis of non-essential amino acids. Exchange of individual amino acids.		
		Exchange of murviqual ammo acids.		
7	Biosynthesis of	Lecture №19 The structure of nucleic acids. Synthesis of		2
/	Diosynthesis of	Declare 31217 The structure of fractice acids, byfinesis of		

	proteins	DNA, RNA, repair	
		<b>Lecture №20</b> Protein biosynthesis. Inhibitors of matrix	2
		biosynthesis. Mechanisms of genetic variability and protein	
		polymorphism. DNA technologies in medicine.	
	Regulation of	<b>Lecture №21</b> Hormonal regulation of water-salt	
	metabolism.	metabolism. Biochemistry of the liver. Inactivation of	2
	Biochemistry of	foreign substances in the body. Metabolism of erythrocytes,	
	individual	heme and formation of pigments.	
	organs and	<b>Lecture №22</b> Biochemistry of connective and bone tissue.	2
8	tissues	Regulation of the process by hormones and vitamins.	
	(Biochemistry	Collagen, elastin, proteoglycans, their role.	
	of the liver,		
	blood, urine,		
	connective		
	tissue).		
		TOTAL	16h

TOTAL:44h

5.4. Thematic plan of practical classes

№ secti on	disciplines section	Topics of practical classes	Forms of current control *	Numb hours	s per
OII				No3	№4
1.	Proteins and enzymes	PC1 Structural organization of proteins. Amino acids as structural components of proteins.	T,I	2	0.2.
		PC.2 Physico-chemical properties of proteins.	T,I	2	
		PC.3 Features of the functioning of oligomeric proteins. Hemoglobin. Its structure and role.	T,I	2	
		PC.4 Features of enzymatic catalysis. Enzyme specificity.	T,I	2	
		PC.5 Factors affecting the activity of enzymes. Regulation of enzyme activity.	T,I	2	
		PC.6 Classification of enzymes.	T,I	2	
		PC.7 Enzymology. Medicines are enzyme inhibitors. The use of enzymes in medicine	T,I	2	
		PC. 8 Colloquium on topics: «Proteins and enzymes»	T. ST	2	
2.	Vitamins and	PC.9 Some questions of nutritional	T,I	2	

	hormones	biochemistry.			
		PC10. Fat-soluble vitamins A, D, E, K.	T,I	2	
		Ascorbic acid - vitamin C, biorol.		2	
		PC.10 Vitamins, hypovitaminosis and	T,I	2	
		beriberi. Classification of vitamins.			
		PC.11 Water-soluble vitamins, their	T,I		
		coenzymatic role. Hypervitaminosis and		2	
		hypovitaminosis.			
		PC.12 Hormones. general characteristics	T,I	2	
		Π3.13 Hormones of the hypothalamus and	T,I		
		pituitary gland			
		PC.14 Hormones. Adenylate cyclase and	T,I		
		cytosolic mechanisms of action of		2	
		hormones.		_	
		DC 15 Harmones of the papers Action of	TIC		
		PC 15 Hormones of the pancreas. Action of insulin.	1,1. 5.	2	
		insuin.		2	
		DC 16 Hammanas of a stancid natura, their	TI		
		PC.16 Hormones of a steroid nature, their mechanism of action.	T,I		
		inechanish of action.			
				2	
		PC.17 Glucocorticoids	T,I		
				2	
		PC.18 Colloquium on topics: «Vitamins	T. ST.		
		and hormones»	1.51.	2	
3	Bioenergy	PC.19 General path of catabolism.	T,I		
	biooxidation	1 C.17 General path of Catabonism.	1,1		
	Diooxidation				
				2	
				_	
		PC.20 Bioenergy and biooxidation	T,I	2	
		Biochergy and brookladion	1,1	_	
		DC 21 Mitachandrial alegans to the second	TI	2	
		PC.21 Mitochondrial electron transport	T,I	2	
		chain. tissue respiration.			
I	I	L	I .	L	1

		PC.22 Oxidative phosphorylation of ADP. Mitchell's chemiosmotic theory.		2	
			T,I		
		PC.23 Colloquium. Bioenergy and biooxidation	T. ST.	2	
4	Carbohydrate metabolism	<b>PC</b> .24. The main carbohydrates in food. Digestion. Biosynthesis and decay (mobilization). Regulation of glycogen synthesis and breakdown by hormones.		2	
		PC.25 Aerobic and anaerobic breakdown of glucose. Glycolysis. Lactic acid formation.	T,I	2	
		PC. 26 Oxidative decarboxylation of pyruvic acid. Tricarboxylic acid cycle.	T,I	2	
		PC.27. Pentose phosphate pathway for glucose oxidation	T,I	2	
		PC.28. Carbohydrate metabolism.	T,I	2	
		PC.29. Solution of situational problems on the topic Carbohydrate metabolism	T,I	2	
		PC.30 Colloquium on the topic Carbohydrate metabolism TOTAL:	T. ST.	2 <b>60 h</b>	
		TOTAL:		00 11	
5	Lipid metabolism	<b>PC</b> .31. Assimilation of dietary lipids. Transport of lipids by chylomicrons.	T,I		2
		PC.32. TAG mobilization. Process regulation. B-oxidation. Exchange of ketone bodies. Eicosanoids. Biosynthesis of fatty acids and fats. Hormonal regulation.	T,I		2
		PC.33. cholesterol exchange. Hypercholesterolemia. Synthesis of bile acids. biological membranes. lipid peroxidation	T,I		2
		PC.34.Colloquium on topics: «Lipid	T. ST		2
	D 4:	metabolism»			1
6	Protein	PC.35. Digestion of proteins. Trans- and			2

me	etabolism	deamination of amino acids. Exchange of		
		individual amino acids. Neutralization of		
		ammonia in tissues. ornithine cycle.		
		<b>PC</b> .36 Metabolism of complex proteins.	T.	2
		Heme metabolism and pigment		
		formation.		
		Neutralization of bilirubin.		
		PC.37.Colloquium on topics: «Protein	T.	2
		metabolism»»		
7	Biosynthesis of	ПЗ.38 Structure of nucleic acids Synthesis	T.	2
	proteins	of DNA. Replication, repair. Transcription,		
		Operon theory, regulation of RNA		
		synthesis at the level of transcription		
		PC.39. protein biosynthesis. Inhibitors of	Т.	2
		matrix biosynthesis. Mechanisms of genetic		
		variability and protein polymorphism. DNA		
		technologies in medicine. PCR diagnostics		
		PC.40. Colloquium on the topic	T. ST.	2
		«Structure and functions of proteins,		
		nucleic acids, matrix biosynthesis»		
8	Regulation of	PC.41. Hormonal regulation of	T,I	2
	metabolism Liver	carbohydrate, lipid and amino acid		
	biochemistry.	metabolism, regulation of water-salt		
		metabolism. Calcium and phosphate		
		metabolism.		
		<b>PC</b> .41. Biochemistry of the liver.	T,I	2
		Inactivation of foreign substances in the		
		body. Biochemistry of urine. Biochemistry		
		of blood.		
		PC.41. Biochemistry of connective and	T,I	2
		bone tissue. Regulation of the process by		
		hormones and vitamins. Collagen, elastin,		
		proteoglycans, their role.		
		PC.41. Colloquium on topics:	T, ST	2
		«Biochemistry of the liver, blood,		
		connective tissue, urine biochemistry		
		TOTAL	28h.	

TOTAL: 88 h

**5.5.** Laboratory studies

No	Disciplines	Name of laboratory classes	Forms of current	Numb	er of
	дисциплины		control*	hours	s per
				seme	ester
				<b>№</b> 3	<i>№</i> 4
1.	Proteins and	LC. 1 Protein precipitation reactions.	Pr,S	2	
	enzymes	Qualitative reactions to proteins. Benzidine			
		test for hemoglobin.			
		LC. 2 Influence of activators and inhibitors	Pr	2	
		on the activity of salivary enzymes.			
2.	Vitamins and	LC. 3 Quantitative determination of vitamin	Pr	2	

	hormones	C LC. 4 Quantification of rutin in tea. Qualitative reaction to vitamin B <sub>2</sub> .			
3.	Bioenergetics. Biooxidation.	LC. 5 Quantification of ATP  Determination of catalase activity	Pr	2	
4	Carbohydrate metabolism	LC. 6 Determination of salivary amylase activity	Pr,I	2	
		TOTAL:		12h.	
5	Lipid metabolism	LC. 1 Determination of lipase activity  Qualitative and quantitative determination of acetone	Pr,I		2
		LC 2 Quantification of cholesterol	Pr,I		2
6	Protein metabolism	LC. 3 Analysis of gastric juice.  Determination of pathological constituents of gastric juice.  Determination of alanine	Pr,I		2
7	Regulation of metabolism Liver biochemistry.	aminotransferase activity  LC. 4. Determination of phenylpyruvic acid in urine  Determination of total blood serum protein by the biuret method	Pr,I		2
		TOTAL			8h
		TOTAL	20h		

5.6. Educational and methodological support for independent work in the discipline 5.6.1. Independent work of the student in the discipline

$N_{\underline{0}}$	Disciplines	Name of works	Labor	Forms of
$\Pi/\Pi$	section		intensity	control
			(hour)	
1.	Proteins and	The structure and functions of oligomeric proteins on the example of hemoglobin in comparison with myoglobin.	2	I
		Hemoglobinopathies, prevalence in the Republic of Dagestan.	2	P
	enzymes	Regulation of enzyme activity.	2	I
		Enzyme inhibitors and their use as therapeutic drugs.	2	Р
2.	Vitamins and	Ideas about replaceable and irreplaceable components of food.	6	I
	hormones	Genetically modified foods in nutrition.	4	P

	Γ <b>AL</b> :		64 h.	
		of questions; pre-examination individual and group consultations with the teacher.		
	*	material, educational literature); wording	24	
	Exam preparation	studied material (work with lecture		
		Repetition and consolidation of the	<u> </u>	
			20 h.	
	connective ussue	Changes in connective tissue during aging and collagenoses.	2	P
	Biochemistry of connective tissue	•		
	Dio als assistants and	disorders of connective tissue proteins.		P
	municia ucius	Hereditary and acquired metabolic	4	
'	Biosynthesis of nucleic acids	Genetic diseases	4	P
		multimedia presentation		
		treatment		I
5	metabolism	Phenylketonuria: causes, symptoms,	2	
	Protein	Gout: causes, symptoms, treatment	2	P
		multimedia presentation		
		Violation of bilirubin metabolism	2	I
	F 55556 5	multimedia presentation	_	•
	Lipid metabolism	Hypercholesterolemia	2	I
		lipid metabolism disorder	2	P
			44 h.	
		glycolysis	4	P
•	metabolism	presentation	7	I
	Carbohydrate	Tricarboxylic acid cycle multimedia	4	
		Carbohydrates metabolism disorders	2	P
		coenzymes.		Г
		Vitamins - precursors of tissue respiration	4	P
3.	Bioenergy	Tissue respiration inhibitors	4	P
		in the development of pathology.		
		Insulin and glucagon as antagonists. Role		T
		Hormonal regulation of metabolism.	4	
		metabolism.		51
		Prostaglandins and their derivatives, biological role and influence on	4	ST

**5.6.2.** Guidelines for students on mastering the discipline *APPENDIX to the WP* 

## VI. EVALUATION TOOLS FOR CURRENT CONTROL OF PROGRESS AND INTERIM CERTIFICATION ON THE RESULTS OF MASTERING THE DISCIPLINE

6.1. The list of competencies indicating the stages of their formation in the process of mastering the work program of the discipline

No॒	discipline section name	Controlled competency code	Forms of control
Discipli	(module)	(or part of it)	
ne			
section			
1	2	3	4
1.	Proteins and enzymes	GPC-5	I,T, ST, P,Pr
2.	Vitamins and hormones	GPC-5	I,T, ST, P,Pr
3	Биоокисление	GPC-5	I,T, ST, P,Pr
	Bioenergy		
4	Carbohydrate	GPC-5	I,T, ST, P,Pr
	metabolism		
5	Lipid metabolism	GPC-5	I,T, ST, P,Pr
6	Protein metabolism	GPC-5	I,T, ST, P,Pr
7	Protein biosynthesis	GPC-5	I,T, <i>ST</i> , <i>P</i>
8	regulation of	GPC-5	I,T, ST, P,Pr
	metabolism.		
	Biochemistry of		
	individual organs and		
	tissues (liver, blood,		
	urine, connective		
	tissue).		

6.2. Criteria for evaluating the results of mastering the discipline

Metrics «unsatisfactory» (satisfactory» (average level) (high level reached)  Competency code GPC-5	
reached)	el) 
Competency code GPC-5	
The Prince of the second secon	
To know The student does not The student has The student is able The student	
know the main mastered the main to independently independently	
provisions in the content of the material highlight the main singles out the	main
studied material of of the discipline, but provisions in the provisions in the	he
the discipline. has gaps in the studied material. studied material.	al and
Does not know the assimilation of the Knows the main is able to give	a
structure and material that do not aspects of the most brief descripti	on of
biochemical prevent further important the main ideas	of
properties of the main assimilation of the biochemical the studied ma	ıterial
classes of educational material. processes and of the discipling	ne.
biologically Has unsystematized various types of Shows deep	
important knowledge of homeostasis in the knowledge an	d
compounds, the main   biologically important   body; chemical and   understanding	in
metabolic pathways   metabolic pathways in   biological essence   predicting the	
for their biochemistry; of the processes possibility of	
transformation; occurring in a developing	
mechanisms of living organism at pathologies, u	sing
transmission and the molecular and knowledge ab	out
implementation of cellular levels. the biochemic	al
genetic information mechanisms of	f their

	during the synthesis of DNA, RNA, proteins;			development
To be able to	The student does not know how to use illustrations, diagrams as auxiliary material, does not know how to search and draw generalized conclusions	Student has difficulty writing metabolic charts.	The student is able to explain the consequences of violations of biochemical processes in the body	The student is able to consistently explain the physiological nature of the processes, as well as explain the cause of the pathologies of HD processes.
To possess	The student does not have the skills of independent work with educational scientific and reference literature.	The student has little knowledge of static biochemistry The student basically owns the skill of using visual material.	The student has knowledge of everything studied program material, the material sets out consistently allows for minor errors and shortcomings in the reproduction of the studied material.	The student independently selects the main position in the studied material and is able to give a brief description of the main ideas of the studied material. The student shows a deep and complete knowledge of the entire volume of the discipline being studied.

6.3. Evaluation tools for monitoring progress

#### STUDENT'S INTERVIEW ON CONTROL QUESTIONS

Topic of the lesson: General characteristics and properties of enzymes. The chemical nature of enzymes. The structure of enzymes. The mechanism of action of enzymes. Features of enzymatic catalysis

#### **Controlled Competency Codes: GPC-5**

- 1. General characteristics and basic properties of enzymes.
- 2. Evidence of the protein nature of enzymes.
- 3. The specificity of the action of enzymes.
- 4. Types of specificity.
- 5. Organ-specific enzymes.
- 6. Isoenzymes, determination of the isoenzyme spectrum of enzymes in the clinic.

#### ✓ Criteria for assessing the current control of progress

- ✓ «Excellentn»:
- ✓ The student has a deep knowledge of the educational material on the topic of the practical lesson, formulated a complete and correct answer to the questions of the topic of the lesson, in compliance with the logic of the presentation of the material, shows the assimilation of the relationship of the main concepts used in the work, was able to answer all clarifying and additional questions. The student demonstrates knowledge of theoretical and practical material on the topic of the lesson.

#### √ «Good»:

The student showed knowledge of the educational material, mastered the basic literature, was able to answer almost completely all the additional and clarifying questions asked. The student demonstrates knowledge of theoretical and practical material on the topic of the lesson, allowing

minor inaccuracies.

#### ✓ «Satisfactory»:

- ✓ The student as a whole mastered the material of the practical lesson, answered not all clarifying and additional questions. The student finds it difficult to correctly assess the proposed task, gives an incomplete answer, requiring leading questions from the teacher.
- ✓ «Unsatisfactory»:

The student has significant gaps in the knowledge of the main educational material of the practical lesson, did not fully disclose the content of the questions, could not answer clarifying and additional questions.

## TESTING TOPIC: BIOCHEMISTRY OF ENZYMES

#### **GPC-5**

1.	In	n dis	orders	of	what	level	of th	e stru	ctural	organiz	ation o	of the	enzyme,	a	molecular	(genetic)
dis	sea	se ca	ın dev	elop	:											
@	1															
2																
3																
4																
do	om	ains														

2. What level of organization of a protein molecule underlies the formation of enzyme specificity:

2

@ 3

@ 4

domains

3. At what level of the structural organization of a protein do its enzymatic properties begin to manifest?:

1

2

@ 3

@ 4

domains

5. The allosteric center of an enzyme is:

the sequence of amino acids in the polypeptide chain;

- a unique combination of amino acid residues in the enzyme molecule involved in the act of catalysis;
- @ section of the enzyme molecule, which serves to interact with the modifier (effector);
- a combination of several types of subunits in different quantitative proportions
- 6. Attachment to the allosteric center of an effector enzyme causes:
  - @ change 3 (and 4) enzyme structure and active site configuration;

cleavage of peptide bonds;

change in the sequence of amino acids in the polypeptide chain;

hydrolytic cleavage of the polypeptide chain

7. The active site of an enzyme is:

amino acid sequence in a polypeptide chain:

@ a unique combination of amino acid residues in the enzyme molecule involved in the act of catalysis;

section of the enzyme molecule, which serves to interact with the modifier (effector);

a combination of several types of subunits in different quantitative proportions

- 8. Coenzyme (coenzyme) what is it?
- @ connection of small molecular weight, necessary for the action of the enzyme and loosely associated with it;

a protein consisting of several oligomeric subunits;

competitive inhibitor;

- a low molecular weight compound whose interaction with an enzyme causes its inactivation;
- 9. The feedback regulation of enzyme activity is based on:
  - @ allosteric effect;

competitive inhibition;

pH;

change in the primary structure of the enzyme

- 10. The allosteric effect is based on:
  - @ conformational change;

cleavage of peptide bonds;

competitive inhibition;

increase in substrate concentration

#### Criteria for assessing the current control of progress (testing):

- <u>«Excellentn»:</u> 100-90%
- <u>«Good»:</u> 89-70%
- «Satisfactory»: 69-51%
- <u>«Unsatisfactory»:</u> <50%
- **6.4.** Intermediate certification based on the results of mastering the discipline
- **6.4.1**. Exam Semester 3
- **6.4.1.** Interview
- **6.4.3**. Example questions for preparing for the exam.

#### - Proteins and enzymes

- 1. Physiological role and biological functions of proteins. Amino acids. Structural organization of proteins.
- 2. Physico-chemical properties of proteins. Denaturation of proteins, reversibility of denaturation; factors that cause denaturation. Reversible and irreversible protein precipitation reactions
- 3. Classification of proteins. Simple proteins albumins, globulins, histones.
- 4. Complex proteins, their representatives nucleoproteins, chromoproteins. Hemoglobin, chemical nature, structure, role. Derivatives of hemoglobin. Hemoglobinopathies.

#### Vitamins and hormones

- 2. Enzymopathology, molecular diseases, application of enzymes in medicine.
- 3. Enzymodiagnostics. Enzyme therapy. Immobilized enzymes. Understanding liposomes..
- **4.** 20. Nutritional biochemistry. Replaceable and irreplaceable components of food. Hyper-, hypo- and avitaminosis.

#### **Biooxidation Bioenergy**

 General patterns of metabolism in a living organism. The main stages of catabolism of substances and their relationship. Understanding Specific and General Pathways of Catabolism. **10.** Energy-rich compounds as universal energy keepers in biological objects. Features of the structure and properties of macroergic compounds, ATP and its analogues.

#### Carbohydrate metabolism

- Basic animal carbohydrates, their biological role Digestion and absorption of carbohydrates in the gastrointestinal tract. Formulas of glucose, fructose, galactose.
- Biosynthesis and breakdown of glycogen, hormonal regulation of glycogen reservation and mobilization.
- Anaerobic breakdown of glucose (glycolysis). Glycogenolysis. Glycolytic oxidoreduction. Physiological significance of anaerobic oxidation. Be able to write all the reactions of glycolysis.
- Substrate phosphorylation during glycolysis. Energy balance of anaerobic glucose oxidation.

#### - Lipid metabolism

- 1. Biological significance and physiological role of lipids in the body. The most important lipids in human tissues. Classification of lipids. Reserve lipids (fats) and membrane lipids (complex lipids).
- 2. Digestion of lipids in the gastrointestinal tract. Features of digestion of fats. The role of bile. Bile acids in the process of digestion and absorption of lipids. Enzymes involved in the digestion of lipids.
- 3. Absorption of lipids. Resynthesis of lipids in the intestinal wall. Formation of chylomicrons and their characteristics.
- 4. Intracellular lipolysis. Fat depot lipases, mechanisms of their action and activation. Hormonal mechanisms of regulation of the activity of fat depot lipases. Physiological role of reservation and mobilization of fats in adipose tissue.

#### - Protein metabolism

- 1. Complete protein nutrition. nitrogen balance. Digestion of proteins in various parts of the gastrointestinal tract. The role of hydrochloric acid.
- 2. Replaceable and irreplaceable amino acids. Complete protein nutrition.
- 3. Proteinases pepsin, trypsin, chymotrypsin. Proenzymes of proteinases and the mechanism of their transformation into active forms. Endopeptidases. Exopeptidases: carboxypeptidases, aminopeptidases, dipeptidases. Absorption of amino acids.

#### VI. Nucleic acid metabolism

- 10. History of discovery and study of nucleic acids. Functions of nucleic acids. Localization of nucleic acids in the cell.
- 11. 11. Structural features of DNA (Crick-Watson model). Structural organization of DNA and RNA. DNA replication. Stages of DNA synthesis. DNA-dependent DNA polymerases. DNA-binding proteins and DNA-unwinding proteins. DNA damage repair.

#### **Biochemistry of blood**

Importance of blood for the body. The chemical composition of blood. Blood plasma proteins and their physiological role. Characterization of the main protein fractions - albumins, globulins, fibrinogen.

#### Biochemistry of urine.

- The chemical composition of the urine of a healthy person. a) organic substances of urine, sources of their formation. b) inorganic substances of urine, mineral components of urine. Hormonal regulation of diuresis.
- Changes in the properties and composition of urine in diseases. Pathological components of urine - glucose, protein, ketone bodies, blood, pigments. The diagnostic value of their determination.

#### Biochemistry of connective tissue.

- 6.4.4 1. Biochemistry of connective tissue. Features of the chemical composition.
- 6.4.5 2. Collagen: features of the chemical composition, conformation.
- 6.4.6 3. Elastin, structural features and functions. Glycosaminoglycans, their structure and properties: hyaluronic acid, chondroitinsulfuric acid, etc.

#### Exam card example.

## FSBEI HE DSMU Department of General and Biological Chemistry Ministry of Health of Russia Field - medical business

### BIOCHEMISTRY EXAMINATION CARD № 1

- 1. Classification of proteins. Simple proteins albumins, globulins, histones.
- 2. Regulation of enzyme activity, regulatory enzymes. Allosteric regulation of enzyme activity.
- 3. Specific pathways of amino acid metabolism
- 4. Write the reactions of anaerobic breakdown of glucose (glycolysis).

Approved at a meeting of the department, protocol dated 24.06. 2022 No. 11

Head Department of General and Biological Chemistry

Professor

E.R. Nagiyev

Head of Academic Affairs Associate Professor

Magomedova Z.M..

#### FSBEI HE DSMU Ministry of Health of Russia

Department of General and Biological Chemistry Field - medical business

### BIOCHEMISTRY EXAMINATION CARD № 2

- 1. Biooxidation. Structures suitable for dehydrogenation.
- 2. Polyunsaturated fatty acids. The role of prostaglandins and their derivatives.
- 3. Glycosaminoglycans (hyaluronic acid, chondroitinsulfuric acid, heparin) and their role in the body.
- 4. Write the reactions of the tricarboxylic acid Krebs cycle.

Approved at a meeting of the department, protocol dated 24.06. 2022 No. 11 Head Department of General and Biological Chemistry

Head Department of General and Biological Chemistry Professor

E.R. Nagiyev

Head of Academic Affairs Associate Professor

Magomedova Z.M..

Assessment					
Metrics	«unsatisfactory» (minimum not reached)	«satisfactory» (minimum level)	«good» (average level)	«excellentn» (high level)	
		competency code GPC-5	<u> </u>	L	
To know	The student does not know the main provisions in the studied material of the discipline.  Does not know the structure and biochemical properties of the main classes of biologically important compounds, the main metabolic pathways for their transformation; mechanisms of transmission and implementation of genetic information during the synthesis of DNA, RNA, proteins;	The student has mastered the main content of the material of the discipline, but has gaps in the assimilation of the material that do not prevent further assimilation of the educational material. Has unsystematic knowledge of biologically important metabolic pathways in biochemistry;	The student is able to independently highlight the main provisions in the studied material. Knows the main aspects of the most important biochemical processes and various types of homeostasis in the body; chemical and biological essence of the processes occurring in a living organism at the molecular and cellular levels	The student independently singles out the main provisions in the studied material and is able to give a brief description of the main ideas of the studied material of the discipline. Shows deep knowledge and understanding in predicting the possibility of developing pathologies, using knowledge about the biochemical mechanisms of their development	
To be able	The student does not know how to use illustrations, diagrams as auxiliary material, does not know how to search and draw generalized conclusions	The student has difficulty writing metabolic diagrams.	The student is able to explain the consequences of violations of biochemical processes in the body	The student is able to consistently explain the physiological nature of the processes, as well as explain the cause of the pathologies of HD processes.	
To possess	The student does not have the skills of independent work with educational scientific and reference literature.	The student has little knowledge of static biochemistry  The student basically owns the skill of using visual material.	The student has knowledge of everything studied program material, the material is presented consistently, allows minor errors and shortcomings in the	The student independently selects the main position in the studied material and is able to give a brief description of the main ideas of the	

	reproduction of the	studied material.
	studied material.	The student
		shows a deep and
		complete
		knowledge of the
		entire volume of
		the studied
		discipline in part

# **6.** <u>LIST OF BASIC AND ADDITIONAL LITERATURE NECESSARY FOR THE</u> <u>DEVELOPMENT OF DISCIPLINE</u>

#### 6.1. Main literature

#### **Printed Sources:**

№	Editions	The number of
		copies in the library
1	Principless of medical biochemistry: учебник «Основы биохимии»	100
	для иностранных студентов лечебного факультета мед.вузов/	
'	В.В. Давыдов, Е.Р. Грабовецкая; ISBN 978-5-906648-174М.:	
1	ЭКО-Вектор, 2016-552с.	

#### **Electronic sources:**

EDS Medical University (Student Advisor) http://www.studmedlib.ru – access to all 2nd year students of the medical faculty of Dagestan State Medical University

#### 6.2. additional literature

#### **Printed Sources:**

№	Editions	The number of copies in the library
1	Laboratory Manual on Biological Chemistry. for foreign students of Medical Department of Higher Education Institutions: tutorial / Baigildina A. A., Davydov V. V Москва: ГЭОТАР-Медиа, 2019 304 с ISBN 978-5-9704-4971-4.	2
2	Biochemistry with exercises and tasks / ed. by Glukhov A. I., Garin V. V Москва: ГЭОТАР-Медиа, 2020 296 с ISBN 978-5-9704-5317-9.	2
3	Biochemistry of the connective tissue. Biochemistry of mixed saliva. Tutorial Glukhov A., Usay L., Golenchenko V., Gubareva A. ГЭОТАР-Медиа. 2019 ISBN 978-5-9704-4972-1 128 с.	2

7. The list of resources of the information and telecommunication network "Internet": sites

#### 8. INFORMATION TECHNOLOGY

When studying the discipline, a common package of documents of Internet materials is used, which provide ample opportunities for improving university training of students. The standard features of

most programs are the implementation of the didactic principle of visualization in training; their use enables students to apply various methods to solve the educational problem.

Teaching methods using information technology.

The teaching methods using information technology used in the classroom "BIOCHEMISTRY" include:

- computer testing;
- demonstration of multimedia materials, including video films;
- A list of search engines (site mooodle.dgmu.ru).
- A list of encyclopedic sites.

#### 11. MATERIAL AND TECHNICAL SUPPORT

_1	. MATERIAL AND T	ECHNICAL SUPI	PUKT				
N п/	buildings, structures facilities premises	Own or operational control, household management rent sublease, gratuitous use  3	Name of disciplines	Appointment equipped buildings, premises *, territories with indication of area (sq.m.)	Name of special rooms and premises for independent work	Equipment of special rooms and premises for self work	The list of licensed software.  Requisites  confirming the document
	44, Shamil avenue	Oper. control.	Biochemistry	For educational and scientific educational process::  Training laboratories No. 5-60 m2 each  Science Laboratory No. 4 - 60m2.  Reactant's - 40 20m².  Laboratory assistant - 30 m².  Preparatory room- 20 m².	training laboratories №5	Amplifier Yamaha EMX 62, Speakers ASK SA- 112, Acer Projector Roll wall screen (white, matte) Drager Luma 267 * 356  Training tables, reagent kits and chemical glassware  Training tables, fume hoods with water and electricity supply - 2 pcs, a cabinet for storing reagents - 2 pcs, chemical glassware.  Chemical tables, fume hoods with reagent kits and chemical glassware.	The list of software (Win HOME 10 Russian OLP (Sublicense agreement Tr000044429 dated 12/08/15); Kaspersky Edition Security for Business - Standard Russian Edition. 100-149 Node (License agreement No. 1081-2015 dated 10/14/2015); Office ProPlus 2013 RUS OLP NL Acdmc (Agreement No. ДП-026 dated 10.16.13), etc.)

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#### 10. HUMAN RESOURCES

Information about the staffing necessary for the implementation of the educational process in the discipline

The total number of scientific and pedagogical workers who implement the discipline is 1 person.

The total number of bets held by scientific and pedagogical workers who implement the discipline is 0.75.

Nº	Full name of the teacher	Terms of attraction (full-time, internal part-time, external part-time, by agreement	Current position, academic degree / academic title	The list of taught disciplines according to the curriculum	Education (which educational institution of vocational education graduated, year)	Level of education, name of the specialty in the diploma, name of the assigned qualification	The volume of the study load in the discipline (share rate)	Informatic additional profession education,	al	Work experience with an indication of the period of work and position
								By specialt y	In pedago gy and psycho logy	
11	Magomedo va Z.M.	full-time	Associate Professor Candidate of Chemistry	Biochemistry	DSU 1996	Specialty, chemistry, chemist	1,0	2012	2015	From 2010 to 2016 assistant, from 2017 to date, assistant professor

#### 11. ASSESSMENT FUND FOR CURRENT MONITORING

(Appendix)

#### 12. WORK PROGRAM SHEET

Changes to the work program are made on the basis of orders and instructions of the rector, as well as on the basis of decisions on improving the educational and methodological support of the discipline, approved at the appropriate level (decision of the academic council), CCMC and are recorded in the list of changes.

#### Work change registration sheet

Academic year	Date and number of notice of	Requisites protocol	Section, Division	Signature of recording change
20 - 20				
20 - 20				
20 - 20				
20 - 20				